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1. DIESEL ENGINES 2. As in the Otto cycle, the red and blue curves are adiabats, so no heat is exchanged during these steps. The ef?ciency of the Diesel cycle is most easily worked out by calculating the heat input Q. halong the yellow step 2 to 3 and the heat expelled during the green step 4 to 1. The ef?ciency is then e= W Q.

DIESEL ENGINES An Introduction to Thermal Physics
in Schroeder), DH H 2O = 285:83 kJ mol 1 and DH = 890:36 kJ mol 1 (12) This time, the ?nal volume is 1 3 of the initial volume, since the 2 moles of water has condensed out as liquid with negligible volume compared to the gases. Thus the atmosphere does work PDV = RTDn (13) = 8:31 J K 1 (298 K)(2) (14) = 4:953 kJ (15)

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