# Chapter 12 Basics Of Chemistry

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chemistry. • Describe oxidation and reduction (redox) reactions. • Discuss the different forms of matter: elements, compounds, and mixtures. • Explain the difference between solutions, suspensions, and emulsions. • Explain pH and the pH scale.

~~Chapter 12 Basics of Chemistry~~

organic chemistry. The study of substances that contain the element carbon. oxidation. A chemical reaction that combines a substance with oxygen to produce an oxide. oxidation-reduction. Also known as redox; a chemical reaction in which the oxidizing agent is reduced and the reducing agent is oxidized. oxidizing agent.

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Chapter 12 Basics of Chemistry. STUDY. PLAY. Chemistry. the science that deals with the composition, structures, and properties of matter and how matter changes under different conditions. Organic Chemistry. the study of substances that contain the element carbon. Inorganic chemistry.

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A chemical reaction that combines a substance with oxygen to produce an oxide. oxidation-reduction. (aka redox) a chemical reaction in which the oxidizing agent is reduced (by losing oxygen) and the reducing agent is oxidized (by gaining oxygen). oxidizing agent. Substance that releases oxygen.

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A stable physical mixture of 2 or more substances. ALKANOLAMINES. Alkaline substances used to neutralize acids or raise the PH of many hair products. EMULSION. An unstable physical mixture of 2 or more immiscible substances, plus a special ingredient called a "emulsifier".

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