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these – vanilla, onion and clove, can

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be used as olfactory indicators on the basis of your observations? Let us do some more activities to understand the chemical properties of acids and bases. 2.1.2 How do Acids and Bases React with Metals? n Rinse both cloth strips with water and again check their odour.

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## Acids Bases Salts Study

Introduction 0:05:45 difference

between Acids & Bases 0:12:06

Aqueous solution of Acids & bases

0:18:06 Naturally occur...

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Acid + base → salt + water + heat. H 2

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2. Reaction of non-metal oxides with bases. Non-metal oxides are acidic in nature  
Base + Non-metal oxide → salt + water + heat.  
 $2\text{NaOH} + \text{CO}_2 \rightarrow \text{Na}_2\text{CO}_3 + \text{H}_2\text{O}$ . To know more about Properties of Acids and Bases, visit



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here. Water Acids and bases in water

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2 Acids Bases and Salts. 1. Common  
Salt (Sodium Chloride): Sodium  
chloride (NaCl) is also known as

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Common or Table Salt. It is formed after the reaction between sodium ...

2. Bleaching Powder ( $\text{CaOCl}_2$ ):

Bleaching powder is also known as chloride of lime. It is a ...

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salts (handouts) Chapter 19...Acids,  
Bases, and Salts ACID-BASE THEORIES  
Acids and bases are all around us and  
part of our everyday life (ex. bodily  
functions, vinegar, carbonated drinks,  
citrus fruits, car batteries, antacids,

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cleaning products, etc.). Chapter 19  
Acids Bases And Salts Answer Key  
Salts . When acid and base neutralize,  
salts are formed.

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Salts Class 7 MCQs Questions with  
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Question 1. An oxide is acidic and has  
a pungent odour.

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~~Chapter 5: In today's Class 7 Science~~  
Sprint session in Chapter 5, we  
discuss acids bases and salts class 7.  
The aspirants...

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Chapter 19 "Acids, Bases, and Salts".

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Use these activities to study the vocabulary and major concepts presented in this chapter. compounds derived from the reaction of a strong base with a weak acid or a strong acid with a weak base or a solution of a weak base and one of its salts.

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